

 <p>Raffles Girls' School (Secondary)</p>	<p>Card 1.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Ca}_{(s)} + \text{H}_2\text{O}_{(l)} \downarrow \text{Ca}(\text{OH})_{2(aq)} + \text{H}_{2(g)}$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 2.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{H}_{2(g)} + \text{XO}_{2(g)} \rightarrow \text{H}_2\text{O}_{(l)}$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 3.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Al}_{(s)} + \text{XFe}_2\text{O}_3(s) \downarrow \text{Fe}_{(s)} + \text{Al}_2\text{O}_{3(s)}$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 4.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Ba}(\text{NO}_3)_{2(aq)} + \text{Na}_2\text{SO}_{4(aq)} \downarrow \text{XBaSO}_{4(s)} + \text{NaNO}_{3(aq)}$ <p>Chemical Equations</p>

 <p>Raffles Girls' School (Secondary)</p>	<p>Card 5.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{AgNO}_3(\text{aq}) + \text{CaCl}_2(\text{aq}) \downarrow$ $\text{AgCl}(\text{s}) + \text{Ca}(\text{NO}_3)_2(\text{aq})$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 6.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{Cu}(\text{s}) + \text{AgNO}_3(\text{aq}) \downarrow$ $\text{xCu}(\text{NO}_3)_2(\text{aq}) + \text{Ag}(\text{s})$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 7.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{HCl}(\text{aq}) + \text{Zn}(\text{s}) \downarrow$ $\text{xCu}(\text{NO}_3)_2(\text{aq}) + \text{H}_2(\text{g})$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 8.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{HNO}_3(\text{aq}) + \text{CuO}(\text{s}) \downarrow$ $\text{Cu}(\text{NO}_3)_2(\text{aq}) + \text{xH}_2\text{O}(\text{l})$ <p>Chemical Equations</p>

<p>Raffles Girls' School (Secondary)</p> 	<p>Card 9.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{C}_6\text{H}_{12}\text{Cl} + \text{O}_2 \downarrow \text{CO}_2 + \text{H}_2\text{O}$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 10.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Xe} + \text{F}_2 \downarrow \text{XeF}_4$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 11.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Na} + \text{H}_2\text{O} \downarrow \text{NaOH} + \text{H}_2$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 12.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{P}_2\text{O}_5 + \text{H}_2\text{O} \downarrow \text{H}_3\text{PO}_4$ <p>Chemical Equations</p>

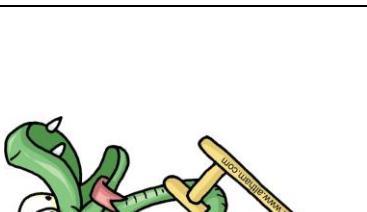
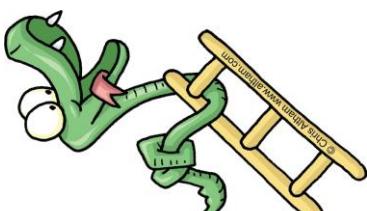
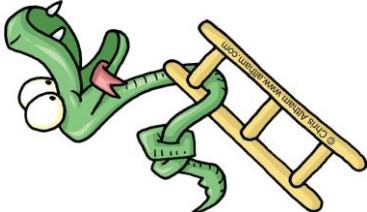
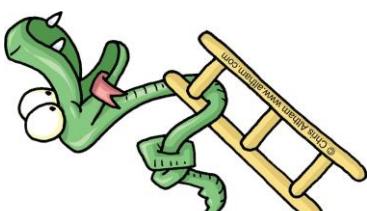
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 13.</p> <p>Balance the chemical equation shown below and then move forward ✕ then move forward ✕ number of spaces on the board:</p> $C_2H_5OH(l) + O_2(g) \downarrow$ $\textcolor{red}{X}CO_{2(g)} + H_2O(l)$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 14.</p> <p>Balance the chemical equation shown below and then move forward ✕ then move forward ✕ number of spaces on the board:</p> $H_2O_{2(l)} \rightarrow \textcolor{red}{X}H_2O_{(l)} + O_{2(g)}$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 15.</p> <p>Balance the chemical equation shown below and then move forward ✕ then move forward ✕ number of spaces on the board:</p> $\textcolor{red}{X}Na_{(s)} + Cl_{2(g)} \rightarrow NaCl_{(s)}$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 16.</p> <p>Balance the chemical equation shown below and then move forward ✕ then move forward ✕ number of spaces on the board:</p> $KNO_{3(s)} \downarrow$ $\textcolor{red}{X}KNO_{2(s)} + O_{2(g)}$ <p>Chemical Equations</p>

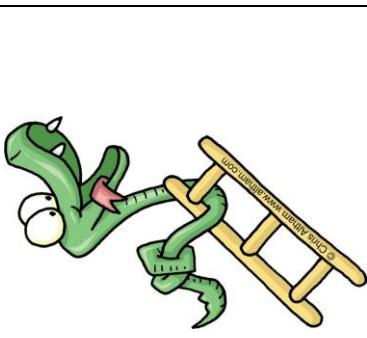
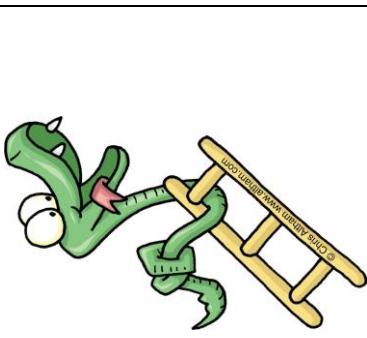
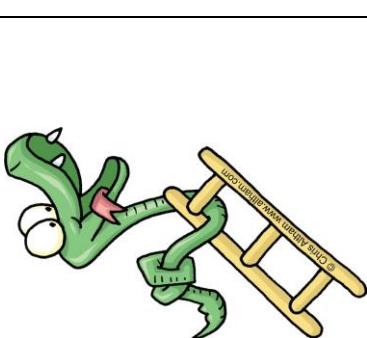
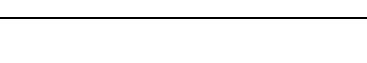
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 17.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{HNO}_3\text{(aq)} + \text{Na}_2\text{CO}_3\text{(s)} \downarrow \text{NaNO}_3\text{(aq)} + \text{H}_2\text{O(l)}$ $+ \text{CO}_2\text{(g)}$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 18.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{XHCl}\text{(aq)} + \text{Mg(s)} \downarrow \text{MgCl}_2\text{(aq)} + \text{H}_2\text{(g)}$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 19.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{CH}_4\text{(g)} + \text{XO}_2\text{(g)} \downarrow \text{CO}_2\text{(g)} + \text{H}_2\text{O(l)}$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 20.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{XHI(g)} \rightarrow \text{H}_2\text{(g)} + \text{I}_2\text{(g)}$ <p>Chemical Equations</p>

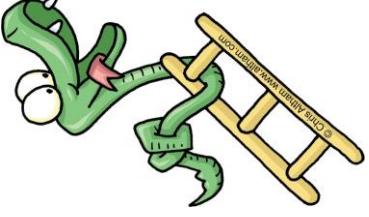
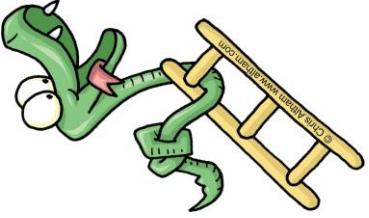
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 21.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$N_2(g) + \textcolor{red}{X}H_{2(g)} \rightarrow NH_{3(g)}$</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 22.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$\textcolor{red}{X}H_2SO_{4(aq)} + Fe_{2O_{3(s)}} \downarrow$</p> <p>$Fe_2(SO_4)_{3(aq)} + H_2O_{(l)}$</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 23.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$Al_{(s)} + \textcolor{red}{X}I_{2(s)} \rightarrow AlI_{3(s)}$</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 24.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$Fe_{2O_{3(s)}} + \textcolor{red}{X}CO_{(g)} \downarrow$</p> <p>$Fe_{(s)} + CO_{2(g)}$</p>

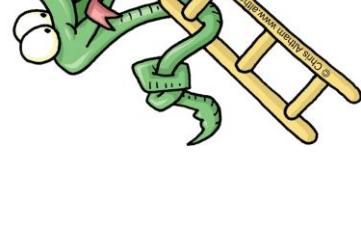
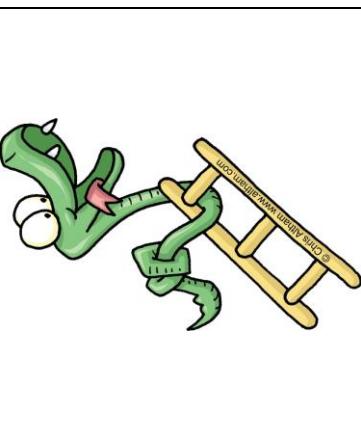
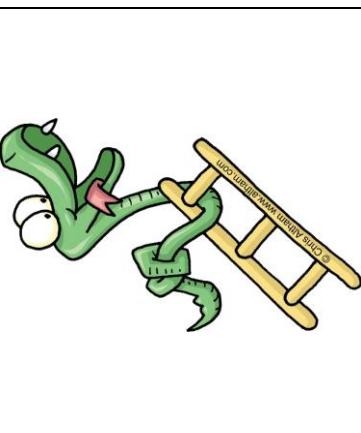
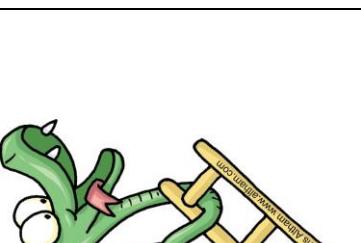
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 25.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$S_{(s)} + \textcolor{red}{X}F_{2(g)} \rightarrow SF_{6(g)}$</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 26.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$H_3PO_{4(aq)} + \textcolor{red}{X}Mg_{(s)} \downarrow Mg_3(PO_4)_2(s) + H_2(g)$</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 27.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$Fe_{(s)} + \textcolor{red}{X}Cl_{2(g)} \rightarrow FeCl_{3(s)}$</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 28.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$B_2O_{3(s)} + \textcolor{red}{X}H_2O_{(l)} \downarrow H_3BO_{3(aq)}$</p>

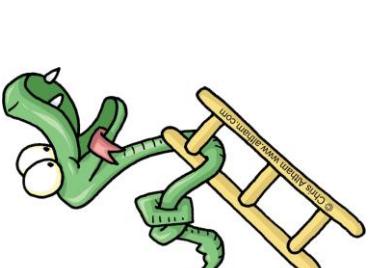
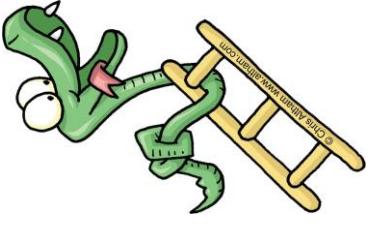
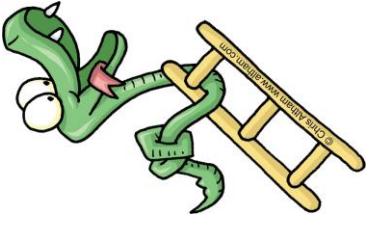
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 29.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{CuO}_{(s)} + \text{NH}_3_{(g)} \downarrow \text{Cu}_{(s)} + \text{H}_2\text{O}_{(l)} + \text{N}_{(g)}$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 30.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{CH}_4_{(g)} + \text{H}_2\text{O}_{(g)} \downarrow \text{CO}_{(g)} + \text{H}_2_{(g)}$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 31.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{XAl}_{(s)} + \text{O}_{2(g)} \rightarrow \text{Al}_2\text{O}_{3(s)}$ <p>Chemical Equations</p>
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 32.</p> <p>Balance the chemical equation shown below and then move forward ✕ number of spaces on the board:</p> $\text{Pb}(\text{NO}_3)_{2(s)} \downarrow \text{PbO}_{(s)} + \text{XNO}_{2(g)} + \text{O}_{(g)}$ <p>Chemical Equations</p>

<p>Card 33.</p> <p>Raffles Girls' School (Secondary)</p>  	<p>Card 33.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{MnO}_{2(s)} + \text{XHCl}_{(aq)} \downarrow \text{MnCl}_{2(aq)} + \text{Cl}_{2(g)} + \text{H}_2\text{O}_{(g)}$
<p>Card 34.</p> <p>Raffles Girls' School (Secondary)</p>  	<p>Card 34.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{CH}_4_{(g)} + \text{XCl}_{2(g)} \downarrow \text{CCl}_{4(l)} + \text{HCl}_{(g)}$
<p>Card 35.</p> <p>Raffles Girls' School (Secondary)</p>  	<p>Card 35.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{C}_2\text{H}_6_{(g)} + \text{O}_{2(g)} \downarrow \text{XCO}_{2(g)} + \text{H}_2\text{O}_{(g)}$
<p>Card 36.</p> <p>Raffles Girls' School (Secondary)</p>  	<p>Card 36.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Cu}_{(s)} + \text{XHNO}_{3(aq)} \downarrow \text{Cu}(\text{NO}_3)_{2(aq)} + \text{H}_2\text{O}_{(l)} + \text{NO}_{2(g)}$

<p>Raffles Girls' School (Secondary)</p>  <p></p>	<p>Card 37.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{FeCO}_{3(s)} + \text{O}_{2(g)} \downarrow \text{Fe}_2\text{O}_{3(s)} + \text{XCO}_{2(g)}$
<p>Raffles Girls' School (Secondary)</p>  <p></p>	<p>Card 38.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{FeCr}_2\text{O}_{4(s)} + \text{XCr}_{(l)} \downarrow \text{Fe}_{(l)} + \text{Cr}_{(l)} + \text{CO}_{(g)}$
<p>Raffles Girls' School (Secondary)</p>  <p></p>	<p>Card 39.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{N}_{2}\text{O}_{5(g)} \rightarrow \text{XNO}_{2(g)} + \text{O}_{2(g)}$
<p>Raffles Girls' School (Secondary)</p>  <p></p>	<p>Card 40.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{H}_{2}\text{S}_{(g)} + \text{H}_{2}\text{SO}_{4(l)} \downarrow \text{S}_{(s)} + \text{XH}_2\text{O}_{(l)}$

 <p>Raffles Girls' School (Secondary)</p> 	<p>Card 41.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $C_3H_8(g) + \textcolor{red}{X}O_{2(g)} \downarrow CO_{2(g)} + H_2O(l)$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p> 	<p>Card 42.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $C_5H_{12}(l) + O_{2(g)} \downarrow \textcolor{red}{X}CO_{2(g)} + H_2O(l)$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p> 	<p>Card 43.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $MnO_4^-(aq) + H^+(aq) \downarrow \textcolor{red}{X}e^- Mn^{2+}_{(aq)} + H_2O(l)$ <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p> 	<p>Card 44.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $V(s) + \textcolor{red}{X}O_{2(g)} \rightarrow V_2O_5(s)$ <p>Chemical Equations</p>

 <p>Raffles Girls' School (Secondary)</p>	 <p>Chemical Equations</p>	<p>Card 45.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $P_{(s)} + \textcolor{red}{X}Cl_{2(g)} \rightarrow PCl_{(s)}$
 <p>Raffles Girls' School (Secondary)</p>	 <p>Chemical Equations</p>	<p>Card 46.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $P_{(s)} + \textcolor{red}{X}O_{2(g)} \rightarrow P_2O_{(s)}$
 <p>Raffles Girls' School (Secondary)</p>	 <p>Chemical Equations</p>	<p>Card 47.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $KIO_{3(aq)} + \textcolor{red}{X}KI_{(aq)} \downarrow + H_2SO_{4(aq)} + I_2_{(aq)} + H_2O_{(l)} + K_2SO_{4(aq)}$
 <p>Raffles Girls' School (Secondary)</p>	 <p>Chemical Equations</p>	<p>Card 48.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $NH_{3(g)} + \textcolor{red}{X}O_{2(g)} \downarrow NO_{(g)} + H_2O_{(l)}$

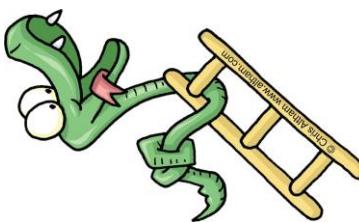
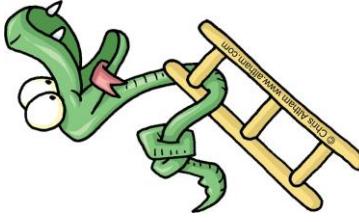
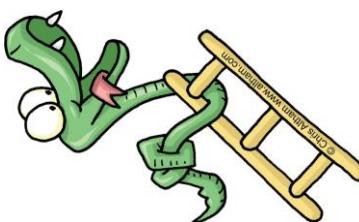
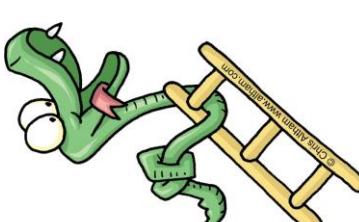
 <p>Raffles Girls' School (Secondary)</p>	 <p>Chemical Equations</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{K}\text{MnO}_{4(\text{s})} + \text{HCl}_{(\text{aq})} \downarrow \text{KCl}_{(\text{aq})} + \text{MnCl}_{2(\text{aq})} + \text{XCl}_{2(\text{g})} + \text{H}_2\text{O}(\text{l})$
 <p>Raffles Girls' School (Secondary)</p>	 <p>Chemical Equations</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Cu}_{2(\text{l})} + \text{Cu}_{2\text{S}(\text{l})} \downarrow \text{XCu}_{(\text{l})} + \text{SO}_{2(\text{g})}$
 <p>Raffles Girls' School (Secondary)</p>	 <p>Chemical Equations</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{XHNO}_{3(\text{aq})} + \text{Fe}_{2\text{O}_3(\text{s})} \downarrow \text{Fe}(\text{NO}_3)_3(\text{aq}) + \text{H}_2\text{O}(\text{l})$

Card 49.

Card 50.

Card 51.

Card 52.

<p>Raffles Girls' School (Secondary)</p> 	<p>Card 53.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{CH}_3\text{COOH}_{(\text{aq})} + \text{Fe}_2\text{O}_3(s) \rightarrow \text{Fe}(\text{CH}_3\text{COO})_3(\text{aq}) + \text{H}_2\text{O}_{(\text{l})}$ <p>Chemical Equations</p> 
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 54.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Cr}_2\text{O}_7^{2-}(\text{aq}) + \text{H}^+_{(\text{aq})} + \text{Xe}^- \rightarrow \text{Cr}^{3+}(\text{aq}) + \text{H}_2\text{O}_{(\text{l})}$ <p>Chemical Equations</p> 
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 55.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Al}_2\text{Cl}_6(s) + \text{XH}_2\text{O}_{(\text{l})} \rightarrow \text{Al}(\text{OH})_3(s) + \text{HCl}_{(\text{g})}$ <p>Chemical Equations</p> 
<p>Raffles Girls' School (Secondary)</p> 	<p>Card 56.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> $\text{Al}_{(\text{s})} + \text{XHCl}_{(\text{aq})} \rightarrow \text{AlCl}_3(\text{aq}) + \text{H}_2\text{G}$ <p>Chemical Equations</p> 

 <p>Raffles Girls' School (Secondary)</p>	<p>Card 57.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$C_6H_{12}O_6(s) + O_2(g) \downarrow XCO_{2(g)} + H_2O(l)$</p> <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 58.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$Al_{(s)} + NaOH_{(aq)} \downarrow XH_2O(l)$</p> <p>$NaAl(OH)_4(aq) + H_2(g)$</p> <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 59.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$Ca_3P_2(s) + XH_2O(l) \downarrow Ca(OH)_2(aq) + PH_3(g)$</p> <p>Chemical Equations</p>
 <p>Raffles Girls' School (Secondary)</p>	<p>Card 60.</p> <p>Balance the chemical equation shown below and then move forward X number of spaces on the board:</p> <p>$C_3H_7OH(l) + O_2(g) \downarrow XCO_{2(g)} + H_2O(l)$</p> <p>Chemical Equations</p>